

Question Paper Code: 94903

B.E. / B.Tech. DEGREE EXAMINATION, DEC 2021

Fourth Semester

Chemical Engineering

19UCH403-Heat Transfer

(Regulation 2019)

Duration: Three hours

Maximum: 100 Marks

PART – A (10 X 2 =20 Marks)

ANSWER ANY TEN QUESTIONS

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| 1. State Fourier's law of heat conduction. | R | CO1 |
| 2. Define specific Heat capacity. | R | CO2 |
| 3. Calculate the rate of heat transfer per unit area through a copper plate 45mm thick. Which one face is maintained at 380°C and other face at 60°C. take thermal conductivity of copper as 370 W/m°C | U | CO1 |
| 4. State Newton's Law of cooling. | R | CO2 |
| 5. What do you mean by Sider-Tate Correction factor? | U | CO1 |
| 6. What is Nusselt number and write its significance, | AN | CO1 |
| 7. What are the properties of radiative heat transfer and explain it. | R | CO2 |
| 8. Define Plancks Law for blackbody radiation | U | CO2 |
| 9. Calculate the rate of heat transfer by radiation from an unlagged steam pipe, 50 mm O.D. at 393 K to air at 293 K. Assume emissivity of 0.9. | AN | CO3 |
| 10. Mention the difference between film wise and drop wise condensation. | AN | CO2 |
| 11. What is film boiling? And what is the risk of attaining film boiling? | AP | CO1 |
| 12. What are the advantages of forced circulation evaporators? | AN | CO5 |
| 13. What is the purpose of chiller in heat exchangers? | R | CO3 |
| 14. What are the two type of tube pitch? Draw a sketch | R | CO4 |

15. What is the purpose of using baffles in a shell & tube heat exchanger?

U CO3

PART - B (5X 16 =80 Marks)

ANSWER ANY THREE QUESTIONS

- 1 A young engineer is asked to design a thermal protection barrier for a sensitive electronic device that might be exposed to irradiation from a high-powered infrared laser. Having learned as a student that a low thermal conductivity material provides good insulating characteristics, the engineer specifies use of a nanostructured aerogel, characterized by a thermal conductivity of $k = 0.005 \text{ W/m K}$, for the protective barrier. The engineer's boss questions the wisdom of selecting the aerogel *because* it has a low thermal conductivity. Consider the sudden laser irradiation of (a) pure aluminum, (b) glass, and (c) aerogel. The laser provides irradiation of $G = 10^6 \text{ W/m}^2$. The absorptivities of the materials are 0.2, 0.9, and 0.8 for the aluminum, glass, and aerogel, respectively, and the initial temperature of the barrier is $T_i = 300 \text{ K}$. Explain why the boss is concerned. *Hint:* All materials experience thermal expansion (or contraction), and local stresses that develop within a material are, to a first approximation, proportional to the local temperature gradient. E CO1
- 2 A heat exchanger is to be designed to heat 1720 kg/h of water from 293 K to 318 K with steam condensing on the outside surface of brass tubes of o.d 25 mm and i.d 22.5 mm and 4 m long. The water velocity is 1.02 m/s, find the number of tubes. $K_{\text{tube material}} = 111.65 \text{ W/(m.K)}$ Weight of steam condensed = 4500 kg/h Latent heat of vaporization of water = 2230 kJ/kg temperature of steam = 383 K steam side film coefficient = $4650 \text{ W/(m}^2 \text{ K)}$. Physical properties of water at mean temperature as given below Density = 995.7 kg/m^3 $C_p = 4.174 \text{ kJ/kg.K}$ Kinematic viscosity $\nu = 0.659 \times 10^{-6} \text{ m}^2/\text{s}$. App CO2
- 3 Calculate the following for an industrial furnace in the form of a black body and emitting radiation at 3000°C. AP CO3
1. Monochromatic emissive power at $\lambda = 1.6 \mu\text{m}$ length
 2. Wavelength at which the emission is maximum.
 3. Maximum emissive power
 4. Total emissive power
 5. Total emissive power of the furnace if it is assumed of a real surface with emissivity equal to 0.9
- 4 A heat exchanger is designed to heat 1720 kg/h of water from 293 K (20 °C) to 318 K (45 °C) with saturated steam condensing on the outside surface of the brass tubes of 25 mm O.D and 22.5 I.D. Tube length is 4 m. Assuming water velocity is being constant at 1.2 m/s. determine the number of tubes required in the heat exchanger. AP CO4
- Data: Thermal conductivity of brass = 460 kJ/(h.m.K)
Latent heat of vaporization of steam = 2230 kJ/kg
Steam side coefficient = $19200 \text{ kJ/(h.m}^2 \text{.K)}$

Physical properties of water at mean fluid temperature are as follows:

Density = 995.7 kg/m^3 , Specific heat = $4.28 \text{ kJ} \cdot (\text{kg} \cdot \text{K})^{-1}$

Thermal conductivity = $2.54 \text{ kJ}/(\text{h} \cdot \text{m} \cdot \text{K})$

Kinematic viscosity = $0.659 \times 10^{-6} \text{ m}^2/\text{s}$

5 Discuss in detail about the design calculations of evaporator

AN CO5