Reg. No.:					

Question Paper Code: 41014

B.E. / B.Tech. DEGREE EXAMINATION, MAY 2022

First Semester

140	JCY104 – ENGINEE	RING CHEMISTR	Y				
(Co	ommon to Civil and M	Iechanical Branches	3)				
	(Regulation						
Duration: Three hours	Answer ALL (Questions.	Maximum: 100 Marks				
	PART A - (10 x 1	= 10 Marks)					
1. Natural rubber is form of polyisoprene							
(a) CiS	(b) trans	(c) PLA	(d) Lexan				
2. Name the monomer present in latex							
(a) butane	(b) ethylene	(c) isoprene	(d) acetylene				
3. Semi-solid lubricant is							
(a) Graphite	(b) MoS ₂	(c) Grease	(d) CNT				
4 is responsible for flash setting of cement							
(a) C_3S	(b) C ₃ A	(c) C_2A	(d) C_2S				
5. An example of cathodic i	nhibitor is						
(a) Hydrazine(c) Benzotriazole	` ′						
6. Name the metal in which	volume of the oxide i	s greater than that c	f metal				
(a) Mg	(b) Cr	(c) Mo	(d)Hg				

7.	Silca is a good	_				
	(a) Adsorbate	(b) Adsorbent	(c) Catalyst	(d) Promoter		
8.]	Multilayer adsorption oc	curs in				
	(a) Physical adsorption(c) Both		(b) Chemical adsorption(d) Ion-exchange adsorption			
9. '	Which transition has low	est energy level ele	ectronic transition?			
	(a) σ - σ*	(b) n - σ*	(c) π –π	(d) n- π *		
10.	. A shift of an absorption	maximum towards	s longer wavelength	is called		
	(a) Blue shift(c) Hyperchromic effect	et	(b) Red shift (d) Hypochrom	ic effect		
		PART - B (5	x 2 = 10 Marks)			
11.	. What is condensation p	oolymerization? Giv	ve an example.			
12.	. What are refractories?	How are they classi	fied?			
13.	. Mention the advantage	s of electroless plat	ing over electroplat	ing.		
14.	. What is Freundlich's a	dsorption isotherm?				
15.	. State Beer- Lamberts la	aw.				
		PART - C (5 x	x 16 = 80 Marks			
16.	. (a) (i) Write the free ra	ndical mechanism f	or the synthesis of I	PVC.	(8)	
	(ii) Differentiate ac	ldition polymerizat	ion from condensat	ion polymerization.	(8)	
			Or			
	(b) (i) Write a detailed	note on fibre reinfe	orced composites.		(8)	
	(ii) Differentiate be	etween thermoplast	ics and thermosets.		(8)	
17.	(a) (i) Explain the ge	neral method for th	e manufacture of re	fractories.	(8)	
	(ii) Describe the pr	ocess of setting and	d hardening of ceme	ent.	(8)	
			Or			
	(b) (i) Explain any fo	ur properties of lub	ricants.		(8)	
	(ii) Explain hydrod	Avnamic lubrication	mechanism		(8)	

	(a)	(1)	Explain the mechanism of electrochemical corrosion.	(8)
		(ii)	What is electroplating? Describe the method of electroplating of gold.	(8)
			Or	
	(b)	(i)	Explain any four basic constituents and functions of paints.	(8)
		(ii)	Give an account of electroless plating of Ni.	(8)
19.	(a)	(i)	Derive an expression for Langmuir adsorption isotherm.	(10)
			Write short notes on autocatalysis.	(6)
			Or	
	(b)	(i)	Explain ion-exchange adsorption in the treatment of water.	(8)
		(ii)	Explain the adsorption theory of catalysis.	(8)
20.	(a)	(i) l	Explain the instrumentation and working of UV-visible spectrophotometer.	(8)
		(ii)	Explain the estimation of Ni by atomic absorption spectroscopy.	(8)
			Or	
	(b)	(i)	Explain the estimation of nickel by atomic absorption spectroscopy.	(8)
		(ii)	Derive Beer-Lambert's law. What are its limitations.	(8)