	Question Paper Code: U4303	
B.E. / B.Tech. DEGREE EXAMINATION, APRIL 2024		
	Professional Elective	
	Electronics and Communication Engineering	
21ECV303- NANO ELECTRONICS		
(Regulations 2021)		
Dura	ation: Three hours Maximum:	100 Marks
Answer ALL Questions		
	PART A - (10 x 2 = 20 Marks)	
1.	Compare nano electronics and nanotechnology	CO1-U
2.	Lithium crystallizes in BCC structure. Calculate the lattice constant, given	
	the atomic weight and density for Lithium are 6.94 gram and 530 Kg/m3	CO2- App
	respectively.	
3.	The reflecting plane of lithium fluoride analyzing crystal has an interplanar distance of $4A^{0}$. Calculate the wavelength of 2^{nd} order differential line which has a glance angle of 30.	CO2- App
4.	Differentiate Physical and Chemical vapour deposition.	CO1-U
5.	Write short notes on the Scanning electron microscopy	CO1-U
6.	Compare X-ray diffraction and Powder diffraction	CO1-U
7.	How nanotechnology is used in fuel cells?	CO1-U
8.	What do you mean by Carbon Clusters? How many carbon atoms are there in a bucky ball?	CO1-U
9.	What is the main advantage of ZnO nanoparticles that make them one of the preferred choices for drug nanocarriers?	CO1-U
10.	What is Bio nano Sensor?	CO1-U

PART – B (5 x 16= 80 Marks)

11. (a) Apply the knowledge of the energy band structure of materials to CO3- App (16) classify the Nanomaterials with neat diagram.

Or

- (b) Apply the knowledge of properties of crystal structure to CO3- App (16) differentiate various structures of Nanomaterials.
- 12. (a) Explain in detail about the Top down Approaches with Grinding CO1-U (16) and Injection molding method.

Or

- (b) Explain the concept of photolithography techniques in the Nano CO1-U (16) materials to fabricate as a Nanofilm.
- 13. (a) Analyze some product form of nano material and explain in detail CO4-Ana (16) how it can be used in the development process of nano devices.

Or

- (b) Explain how photoluminescence properties of nano materials are CO4-Ana (16) used in quantum dots.
- 14. (a) Explain the vibrational and mechanical properties of the CNT by CO1-U (16) applying scanning electron microscopy.

Or

- (b) Explain about the CO1-U (16)
 (i) Carbon Clusters
 (ii) Alkali Doped C60
 (iii) Superconductivity in C60
- 15. (a) Illustrate the principle of Molecular and Supermolecular switches CO1-U (16) with suitable diagram and compare its results.

Or

(b) Explain the concept of nano-biosensors and smart dust. CO1-U (16)