A		Reg. No. :											
		Ouestion Pape	er C	ode	: 53	B B0:	3						
	DE / D Tash DECREE EXAMINATION NOV 2018												
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	I nira Semester												
Biomedical Engineering													
15UBM303 - BIOCHEMISTRY													
(Regulation 2015)													
Dur	Duration: Three hours Maximum: 100 Marks												
	Answer ALL Questions												
		PART A - (10	x 1 =	= 10	Mar	ks)							
1.	Which one the followin principle?	ng property is based on	the a	affinit	ty ch	roma	togra	phy				CC)1- U
	(a) Non-covalent binding of proteins to other molecule												
	(b) Covalent binding of proteins to other molecule												
	(c) Non-covalent binding of lipids to other molecule												
	(d) Covalent binding of lipids to other molecule												
2.	The hydrolysis product	s of sucrose are										CC)1- R
	(a) Maltose	(b) Glucose only	(c) Gl	ucos	e & F	ructo	se	((d) F1	ucto	se on	ıly
3.	Which law is involved in colorimeter? CO2-)2- U					
	(a) Lamberts law		(b) Tl	herm	odyn	amic	s law					
	(c) Chargaff's law		(d) W	atso	n-cric	k mc	del					
4.	Trimalmitin belongs to	the category of										CC)2- R
	(a) Proteins	(b) Lipids	(c) En	zyme	es			(d)	Non	e of	the a	bove
5.	Which of the following	Vhich of the following bases is not present in DNA? CO3- R)3- R					
	(a) Adenine	(b) Guanine	(c) Ur	acil				(d)	Cyte	osine		

6.	Ribose and deoxyribose differ in structure around a single carbon namely,								
	(a) C	21	(b) C2	(c) C3	(d) C4				
7.	Nuc	eic acids are polym	ers of			CO4- R			
	(a) Nucleotides			(b) Nucleosides					
	(c) N	luclei of heavy meta	als	(d) Proteins					
8.	One of the following bonds in protein structure that are not broken on denaturation?					CO4- U			
	(a) 1	Hydrogen bonds	(b) Peptide bonds	(c) Ionic bonds	(d) Disulfide	sulfide bonds			
9.	Which one of the nitrogenous base present in lecithin?								
	(a) C	Choline	oline (b) Ethanolamine (c) Inositol						
10.	Name a non-protein compound that bring about catalysis in biological system.					CO5 -R			
	(a) [(a) DNA (b) RNA (c) Lipids			(d) Carbohydrates				
	PART - B (5 x 2 = 10 Marks)								
11.	List out the types of tools used in Biochemistry.								
12.	What is electrophoresis?								
13.	What are reducing sugar? Give example.								
14.	. State the difference between DNA and RNA. CO4-								
15.	. Mention the common property of lipids. CO								
PART – C (5 x 16= 80Marks)									
16.	(a)	Explain the stabili laws involved in the	zing forces involved in nermodynamics.	molecules and what are the	e CO1-U	(16)			
	Or								
	(b)	Explain in detail using different too	how the molecules are ls in biochemistry.	separated from one another	CO1- U	(16)			
17.	(a)	What is the basic the mass of the mo	principle of mass spectrolecule.	ometry and how to calculate	e CO2-U	(16)			
Or									
	(b)	Explain the metal	oolism of Nitrogen con	ntaining compounds through	CO2-U	(16)			

Nitrogen fixation and photosynthesis.

18.	(a)	Describe the glycolytic pathway and calculate the yield number of ATP molecules per molecule of glucose degraded.	CO3- U	(16)
		Or		
	(b)	Explain the Glycogenesis pathway and its regulation.	CO3- U	(16)
19.	(a)	Discuss the properties of amino acids and Explain in detail how DNA act as a genetic material.	CO4- U	(16)
		Or		
	(b)	Describe the Watson and Crick Model Structure of DNA.	CO4- U	(16)
20.	(a)	Explain the different types of enzymes and describe any one method used to measure enzyme activity. Or	CO5- U	(16)
	(b)	Discuss the general classification of lipids and explain how phospholipids is present within lipids and give its structure.	CO5- U	(16)