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Question Paper Code: 31555

B.E. / B.Tech. DEGREE EXAMINATION, MAY 2016

Fifth Semester

Electronics and Instrumentation Engineering

01UEI505 – ANALYTICAL INSTRUMENTS

(Common to Instrumentation and Control Engineering)

(Regulation 2013)

Duration: Three hours

Maximum: 100 Marks

Answer ALL Questions

PART A - (10 x 2 = 20 Marks)

1. State Beer's Law.
2. What are the techniques used for handling solid samples in IR spectroscopy?
3. What are the different chromatography techniques?
4. What are the requirements for a pumping system in HPLC?
5. What is the use of chemiluminescence method?
6. Write the types of H₂S analyzer.
7. What are the properties of reference electrode?
8. Write a note on silicon analyzer.
9. Write the types of NMR spectrometer.
10. What are the applications of mass spectrometry?

PART - B (5 x 16 = 80 Marks)

11. (a) With a neat sketch explain the principle and operation of instrumentation of atomic absorption spectrophotometer. (16)

Or

- (b) Explain the working principle of FTIR spectrophotometers with neat sketch. (16)

12. (a) With a neat diagram explain the principle of separation in gas chromatography and write the advantages and disadvantages of gas chromatography. (16)

Or

- (b) (i) Describe the operation and working of HPLC with neat diagram. (12)

- (ii) What are the advantages of HPLC over gas chromatography? (4)

13. (a) (i) Explain briefly about the principle of infrared analyzers. (6)

- (ii) Explain the types of IR analyzers and its application with neat sketch. (10)

Or

- (b) (i) Explain with neat diagram of thermal conductivity analysers. (6)

- (ii) Describe with neat diagram about the estimation of carbon monoxide in a gas. (10)

14. (a) Describe the constructional details of reference electrode used for different pH ranges and write the precautions in the use of reference electrodes. (16)

Or

- (b) Explain with neat sketch of dissolved oxygen analyzer. (16)

15. (a) Describe about the construction and working of basic ESR spectrometer with neat block diagram. (16)

Or

- (b) Explain the principle of operation of a mass spectrometer. (16)