Reg. No.:					

Question Paper Code: 41014

B.E. / B.Tech. DEGREE EXAMINATION, MAY 2017

First Semester

140	UCY104 – ENGINEE	ERING CHEMISTR	Y			
(Common to Civil and Mechanical Branches)						
(Regulation 2014)						
Duration: Three hours	Answer ALL	Questions.	Maximum: 100 Marks			
	PART A - (10 x	1 = 10 Marks)				
1. Natural rubber is	form of polyisopre	ene				
(a) CiS	(b) trans	(c) PLA	(d) Lexan			
2. Name the monomer present in latex						
(a) butane	(b) ethylene	(c) isoprene	(d) acetylene			
3. Semi-solid lubricant is						
(a) Graphite	(b) MoS ₂	(c) Grease	(d) CNT			
4 is respons	sible for flash setting	of cement				
(a) C ₃ S	(b) C ₃ A	(c) C ₂ A	(d) C_2S			
5. An example of cathodic i	nhibitor is					
(a) Hydrazine(c) Benzotriazole	(b) Calcium sulphat(d) Calcium carbona					
6. Name the metal in which	volume of the oxide	is greater than that o	of metal			
(a) Mg	(b) Cr	(c) Mo	(d)Hg			

7.	Silca is a good	_			
	(a) Adsorbate	(b) Adsorbent	(c) Catalyst	(d) Promoter	
8.	Multilayer adsorption o	ecurs in			
	(a) Physical adsorption(c) Both	1	(b) Chemical ad (d) Ion-exchang	-	
9.	Which transition has lov	vest energy level ele	ectronic transition?		
	(a) σ - σ*	(b) n - σ*	(c) π –π	(d) n- π *	
10	. A shift of an absorption	n maximum towards	s longer wavelength	is called	
	(a) Blue shift(c) Hyperchromic effe	ct	(b) Red shift(d) Hypochromi	c effect	
		PART - B (5	x 2 = 10 Marks)		
11	. What is condensation	polymerization? Giv	ve an example.		
12	. What are refractories?	How are they classi	fied?		
13	. Mention the advantage	es of electroless plat	ing over electroplat	ing.	
14	. What is Freundlich's a	dsorption isotherm?)		
15	. State Beer- Lamberts 1	aw.			
		PART - C (5 x	x 16 = 80 Marks)		
16	. (a) (i) Write the free r	adical mechanism fo	or the synthesis of I	PVC.	(8)
	(ii) Differentiate a	ddition polymerizati	ion from condensati	ion polymerization.	(8)
			Or		
	(b) (i) Write a detailed	l note on fibre reinfo	orced composites.		(8)
	(ii) Differentiate b	etween thermoplasti	ics and thermosets.		(8)
17	. (a) (i) Explain the ge	eneral method for the	e manufacture of re	fractories.	(8)
	(ii) Describe the p	rocess of setting and	I hardening of ceme	ent.	(8)
			Or		
	(b) (i) Explain any fo	our properties of lub	ricants.		(8)
	(ii) Explain hydro	dynamic lubrication	mechanism.		(8)

18.	(a)	(i)	Explain the mechanism of electrochemical corrosion.	(8)
		(ii)	What is electroplating? Describe the method of electroplating of gold.	(8)
			Or	
	(b)	(i)	Explain any four basic constituents and functions of paints.	(8)
		(ii)	Give an account of electroless plating of Ni.	(8)
19.	(a)	(i)	Derive an expression for Langmuir adsorption isotherm.	(10)
		(ii)	Write short notes on autocatalysis.	(6)
			Or	
	(b)	(i)	Explain ion-exchange adsorption in the treatment of water.	(8)
		(ii)	Explain the adsorption theory of catalysis.	(8)
20.	(a)	(i) I	Explain the instrumentation and working of UV-visible spectrophotometer.	(8)
		(ii)	Explain the estimation of Ni by atomic absorption spectroscopy.	(8)
			Or	
	(b)	(i)	Explain the estimation of nickel by atomic absorption spectroscopy.	(8)
		(ii)	Derive Beer-Lambert's law. What are its limitations.	(8)