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Question Paper Code: 21425

B.E./B.Tech. DEGREE EXAMINATION, MAY/JUNE 2013.

Fifth Semester

Electronics and Instrumentation Engineering

EI 2302/EI 52/10133 EI 505 — ANALYTICAL INSTRUMENTS

(Common to Instrumentation and Control Engineering)

(Regulation 2008/2010)

Time: Three hours

Maximum: 100 marks

Answer ALL questions.

PART A — $(10 \times 2 = 20 \text{ marks})$

- 1. What is total internal reflection in spectrometry?
- 2. State Beer-Lambert's law?
- 3. List the types of detectors used in gas chromatography.
- 4. What is HPLC?
- 5. Brief the principle of dust monitor.
- 6. What is thermal conductivity?
- 7. Define PH of a solution.
- 8. List the demerits of glass electrode.
- 9. Brief the principle of scanning electron microscope.
- 10. What is mass spectrometer?

PART B - (5 × 16 = 80 marks)

11. (a) With a neat sketch, explain the instrumentation setup and working principle of IR spectrometer.

Or

(b) With a neat diagram, explain the instrumentation setup of Atomic absorption spectrophotometer.

12. (a) With necessary diagrams, explain the working principle of HPLC. (High pressure liquid chromatography)

Or

- (b) With suitable diagrams, explain the various sampling techniques in Gas Chromatography.
- 13. (a) With a neat sketch, explain the construction and working principle of parametric oxygen analyzer.

Or

- (b) (i) With suitable diagrams, explain the construction of smoke meter and its working principle.
 - (ii) Explain the working principle of NO₂ analyzer.
- 14. (a) With neat diagram explain the working principle of dissolved oxygen analyzer.

Or

- (b) With neat diagram, explain the set up of sodium analyzer used in industry.
- 15. (a) With a neat sketch, explain the NMR spectrometry and its instrumentation set up.

Or

(b) With a neat sketch, explain the instrumentation set up of Mass spectrometer and its function.