Question Paper Code: 35505

B.E. / B.Tech. DEGREE EXAMINATION, APRIL 2019

Fifth Semester

Electronics and Instrumentation Engineering

(Common to Instrumentation and Control Engineering)

01UEI505 - ANALYTICAL INSTRUMENTS

(Regulation 2013)

Duration: Three hours

Maximum: 100 Marks

Answer ALL Questions

PART A - (10 x 2 = 20 Marks)

- 1. State Beer's Law.
- 2. What are the techniques used for handling solid samples in IR spectroscopy?
- 3. What are the different chromatography techniques?
- 4. What are the requirements for a pumping system in HPLC?
- 5. List the method used for measuring oxygen in flue gas.
- 6. Describe the principle of thermal conductivity analyzer.
- 7. List is the use of black in silica analyzer.
- 8. Write a note on silicon analyzer.
- 9. Write the types of NMR spectrometer.
- 10. What are the applications of mass spectrometry?

PART - B (5 x 16 = 80 Marks)

11. (a) With a neat sketch explain the principle and operation of instrumentation of atomic absorption spectrophotometer. (16)

Or

- (b) Explain the single beam and double beam spectrophotometers. (16)
- 12. (a) With a neat diagram explain the principle of separation in gas chromatography and write the advantages and disadvantages of gas chromatography. (16)

Or

- (b) Analyze the components and operation of the high performance liquid chromatography. (16)
- 13. (a) Explain the function of Hay's magneto strictive analyzer used for measurement of oxygen. (16)

Or

- (b) Explain the measurement of sulphur dioxide and carbon monoxide. (16)
- 14. (a) Describe the constructional details of reference electrode used for different pH ranges and write the precautions in the use of reference electrodes. (16)

Or

- (b) Examine the working principle of sodium analyzer. (16)
- 15. (a) (i) Explain in detail about the NMR Spectrometer. (8)
 - (ii) Illustrate the working of Quadrapole Mass Spectrometer. (8)

Or

(b) Explain the principle of operation of a mass spectrometer. (16)